

SYNTHESIS OF 3-CYANO- Δ^2 - AND - Δ^3 -DIHYDRO-PYRAN AND -THIOPYRAN DERIVATIVES

F. J. LOPEZ APARICIO, F. SANTOYO GONZALEZ, P. GARCIA MENDOZA, AND F. HERNANDEZ MATEO
Department of Organic Chemistry, Faculty of Sciences, University of Granada, Granada (Spain)
(Received March 7th, 1986; accepted for publication, May 28th, 1986)

ABSTRACT

Treatment of *trans,trans*-3,5-diacyloxy-*r*-4-*tert*-butoxycarbonyl-4-cyano-tetrahydropyran (**1a-b**) and -tetrahydrothiopyran (**1c-d**), 5-(2,4-di-*O*-acyl-3-*tert*-butoxycarbonyl-3-cyano-3-deoxy- β -D-xylo-pentopyranosyl)-3-ethoxycarbonyl-2-methylfuran (**2a-b**) and -3-acetyl-2-methylfuran (**2c-d**), and methyl 2,4,6-tri-*O*-acetyl-3-*tert*-butoxycarbonyl-3-cyano-3-deoxy- α -D-*gluco*-hexopyranoside (**3**) with toluene-*p*-sulfonic acid in acetic anhydride gave the corresponding branched unsaturated compounds, 5-acyl-4-cyano-5,6-dihydro-2*H*-pyran (**4a-b**) and -thiopyran (**4c-d**), 5-(4-*O*-acyl-3-cyano-2,3-dideoxy- β -D-*glycero*-pent-2-enopyranosyl)-furans (**5a-d**) (minor), 5-(2-*O*-acyl-3-cyano-3,4-dideoxy- β -D-*glycero*-pent-3-enopyranosyl)furans (**6a-d**) (major), and methyl 2,6-di-*O*-acetyl-3-cyano-3,4-dideoxy- α -D-*erythro*-hex-3-enopyranoside (**7**).

INTRODUCTION

We have reported on the synthesis and reactivity of 3-deoxy-C-glycosyl derivatives and 3-deoxyglycosides branched at C-3 by the reaction of 3-hetero-1,5-dialdehydes with active methylene compounds¹⁻³. When *tert*-butyl cyanoacetate is used, products can be transformed into branched, unsaturated C- or O-glycosyl derivatives by acid-catalysed loss of the *tert*-butoxycarbonyl group.

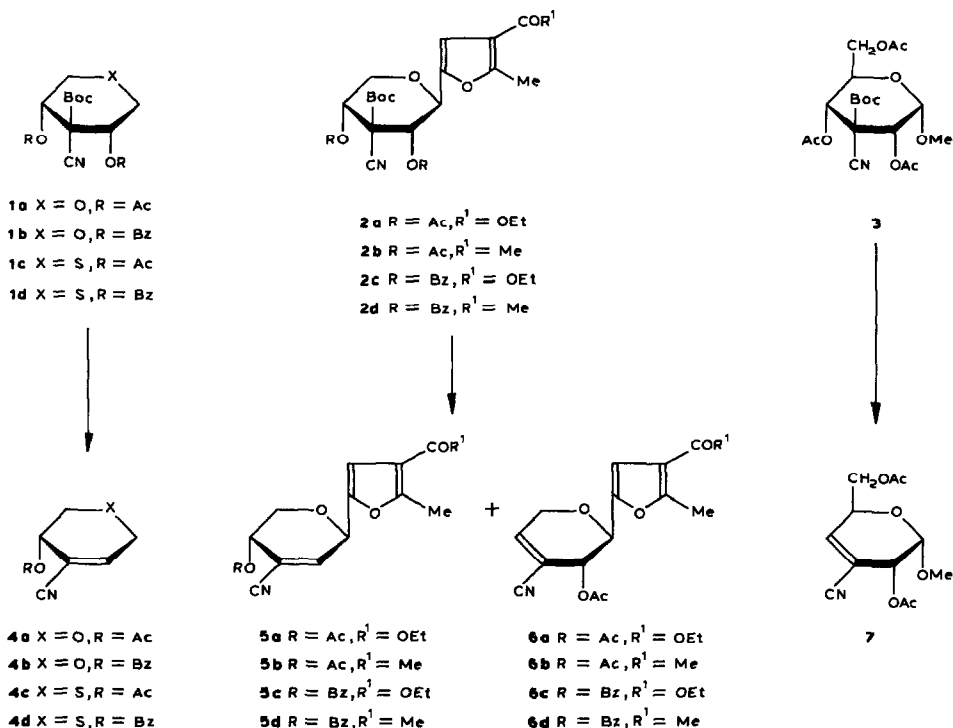
This reaction has now been applied to **1a-d**, **2a-d**, and **3** obtained³ by the reaction of *tert*-butyl cyanoacetate variously with diglycolaldehyde, thiodiglycolaldehyde, (2*S*)-2-(3-ethoxycarbonyl-2-methyl-5-furyl)-3,5-dihydroxy-1,4-dioxane, (2*S*,3*R*,5*S*)-2-(3-acetyl-2-methyl-5-furyl)-3,5-dihydroxy-1,4-dioxane, and α -(*S*)-methoxy- α' -(*R*)-hydroxymethyldiglycolaldehyde.

RESULTS AND DISCUSSION

The reaction was carried out in boiling acetic anhydride, using toluene-*p*-sulfonic acid as catalyst, and the products were isolated by column chromatography.

The pyrans **1a,b** and the thiopyrans **1c,d** gave the 5-acyl-4-cyano-5,6-dihydro-

2*H*-pyrans (**4a,b**) and -thiopyrans (**4c,d**), respectively. With the *C*-pyranosyl derivatives **2a-d**, the major products were 5-(2-*O*-acyl-3-cyano-3,4-dideoxy- β -D-glycero-pent-3-enopyranosyl)furans (**6a-d**), and the minor products were 5-(4-*O*-acyl-3-cyano-2,3-dideoxy- β -D-glycero-pent-2-enopyranosyl)furans (**5a-d**). The sole product from **3** was methyl 2,6-di-*O*-acetyl-3-cyano-3,4-dideoxy- α -D-erythro-hex-3-enopyranoside (**7**).



The structures of **4a-d**, **5a-d**, **6a-d**, and **7** were established on the basis of elemental analyses and spectroscopic data. Thus, **4a-d**, **5a-d**, **6a-d**, and **7** show characteristic i.r. bands⁴ for alkene, ester, and α,β -unsaturated nitrile groups. The mass spectra of **4a,c,d**, **6b,d**, and **7** contained stronger peaks for ($M^+ + 1$) ions than for molecular ions, as has been reported⁵.

The half-chair form is assumed, as reported⁶⁻¹¹ for other dihydropyran derivatives. Compounds **4a,b** had $J_{5,6} + J_{5,6'}$ 6.1 and 6.6 Hz, respectively (see Table II), which accord¹² with an almost exclusive 6H_O conformation (**8A**). This is also in agreement¹³ with the $J_{3,5}$ values of 0.7 Hz. The allylic effect⁶ favours conformers 6H_O . In contrast, **4c,d** had $J_{5,6} + J_{5,6'}$ 12.0 and 11.5 Hz, respectively (see Table II), suggesting¹² a $\sim 2:1$ equilibrium of the half-chair forms 5H_6 (**9A**) and 6H_5 (**9B**), probably due to the "hockey-stick" effect¹⁴. Compounds **5a-d** had $J_{4,5} + J_{4,5'}$ 6.3–6.5 Hz, and $J_{2,4} < 0.7$ Hz (see Table IV) in agreement with the preferred

$^5H_o(D)$ conformers (**8A**). This conformation is stabilised by the allylic effect. Compounds **6a-d** had $J_{4,5} + J_{4,5'}$ 5.0–5.6, $J_{2,4}$ values 1.0–1.5, $J_{2,5}$ 2.4, and $J_{1,2}$ 6.0–6.5 Hz (see Table IV), indicating a preference for the $^0H_1(D)$ conformation (**9A**).

The homoallylic $J_{2,5}$ value of 3.5 Hz (both protons quasi-diaxial)^{12,15} for **7**, the allylic $J_{2,4}$ value of 2.3 Hz (quasi-axial allylic proton), and the $J_{1,2}$ value of 4.0 Hz

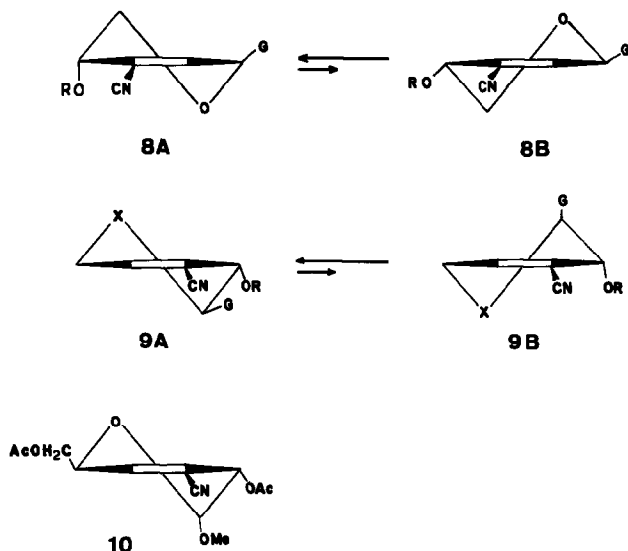


TABLE I

1H -N.M.R. CHEMICAL SHIFTS (δ p.p.m.)^a FOR **4a-d**

Compound	H-2	H-2'	H-3	H-5	H-6	H-6'	Others
4a^b	4.30ddd	4.16dt	6.90ddd	5.21m	3.86dd	3.75dd	2.09 (s, 3 H, Ac)
4b^b	4.40ddd	4.25dt	6.98ddd	5.49m	4.02dd	3.92dd	8.10 and 7.5 (2 m, 5 H, Ph)
4c^c	$\leftarrow 3.30m \rightarrow$		6.95dt	5.50m	3.00dd	2.80dd	2.17 (s, 3 H, Ac)
4d^c	$\leftarrow 3.60-3.10m \rightarrow$		6.95dt	5.70m	3.10dd	2.80dd	8.10 and 7.5 (2 m, 5 H, Ph)

^aFor solutions in CDCl₃ (internal Me₄Si). ^b200 MHz. ^c80 MHz.

TABLE II

PROTON-PROTON COUPLING CONSTANTS^a (Hz) FOR **4a-d**

Compound	$J_{2,2'}$	$J_{2,3}$	$J_{2',3}$	$J_{2,5}$	$J_{2',5}$	$J_{3,5}$	$J_{5,6}$	$J_{5,6'}$	$J_{6,6'}$
4a	19.3	3.4	2.4	1.3	2.4	0.7	2.9	3.2	12.7
4b	19.3	3.3	2.4	1.4	2.4	0.7	3.2	3.4	12.6
4c	19.0	4.0	4.0	2.5	2.0	1.5	5.0	7.0	13.5
4d	19.0	3.5	3.5	2.5	1.5	1.6	4.5	7.0	13.5

^aA series decoupling experiments enabled assignments to be made.

TABLE III

¹H-N.M.R. CHEMICAL SHIFTS (δ , p.p.m.)^a FOR **5a-d**, **6a-d**, AND **7**

Compound	H-2	H-1	H-4	H-5	H-5'	Others
5a^b	6.95bd	$\leftarrow$ 5.35m $\rightarrow$		4.0dd	3.8dd	6.57 (s, 1 H, furan H-4), 4.25 (q, 2 H, J 7.0 Hz, CH ₃ CH ₂ O), 2.6 (s, 3 H, furan Me-2), 2.17 (s, 3 H, Ac), and 1.35 (t, 3 H, J 7.0 Hz, CH ₃ CH ₂ O)
5b^b	6.90dd	$\leftarrow$ 5.30-5.15m $\rightarrow$		3.95dd	3.85dd	6.57 (s, 1 H, furan H-4), 2.60 (s, 3 H, furan Me-2), 2.40 (s, 3 H, furan Ac-3), and 2.17 (s, 3 H, Ac)
5c^b	7.0dd	5.45d	5.55m	4.2dd	3.95dd	8.10 and 7.50 (2 m, 5 H, Ph), 6.62 (s, 1 H, furan H-4), 4.30 (q, 2 H, J 7.0 Hz, CH ₃ CH ₂ O), 2.60 (s, 3 H, furan Me-2), and 1.35 (t, 3 H, J 7.0 Hz, CH ₃ CH ₂ O)
5d^b	7.0bd	5.45d	5.55m	4.15dd	3.95dd	8.10 and 7.50 (2 m, 5 H, Ph), 6.7 (s, 1 H, furan H-4), 2.60 (s, 3 H, furan Me-2), and 2.40 (s, 3 H, furan Ac-3)
6a^c	5.70m	4.63d	6.88dt	4.42dt	4.30dt	6.61 (s, 1 H, furan H-4), 4.25 (q, 2 H, J 7.0 Hz, CH ₃ CH ₂ O), 2.55 (s, 3 H, furan Me-2), 2.1 (s, 3 H, Ac), and 1.32 (t, 3 H, J 7.0 Hz, CH ₃ CH ₂ O)
6b^c	5.67m	4.54d	6.84dt	4.38dt	4.25dt	6.55 (s, 1 H, furan H-4), 2.60 (s, 3 H, furan Me-2), 2.40 (s, 3 H, furan Ac-3), and 2.12 (s, 3 H, Ac)
6c^b	6.0m	4.86d	6.95dt	$\leftarrow$ 4.40m $\rightarrow$		8.10 and 7.50 (2 m, 5 H, Ph), 6.67 (s, 1 H, furan H-4), 4.3 (q, 2 H, J 7.0 Hz, CH ₃ CH ₂ O), 2.55 (s, 3 H, furan Me-2), and 1.32 (t, 3 H, J 7.0 Hz, CH ₃ CH ₂ O)
6d^b	6.0m	4.80d	6.95dt	$\leftarrow$ 4.40m $\rightarrow$		8.10 and 7.50 (2 m, 5 H, Ph), 6.65 (s, 1 H, furan H-4), 2.55 (s, 3 H, furan Me-2), and 2.35 (s, 3 H, furan Ac-3)
7^c	5.41-5.36td	5.13d	6.71pscdot	4.47m	—	4.3 (dd, 1 H, J 11.5, 5.0 Hz, H-6), 4.22 (dd, 1 H, J 11.5, 5.0 Hz, H-6'), 3.49 (s, 3 H, MeO), 2.20 and 2.11 (2 s, 6 H, 2 Ac)

^aFor solutions in CDCl₃ (internal Me₄Si). ^b80 MHz. ^c200 MHz.

TABLE IV

PROTON-PROTON COUPLING CONSTANTS^a (Hz) FOR COMPOUNDS **5a-d**, **6a-d**, AND **7**

<i>Compound</i>	$J_{1,2}$	$J_{2,4}$	$J_{4,5}$	$J_{4,5'}$	$J_{5,5'}$	$J_{2,5}$
5a	3.0	<0.7	3.0	3.3	10.0	—
5b	3.5	<0.7	3.0	3.4	13.0	—
5c	3.5	<0.7	3.0	3.5	9.0	—
5d	3.0	<0.7	3.0	3.5	12.5	—
6a	6.0	1.3	2.8	2.8	19.5	2.4
6b	6.5	1.4	2.7	2.7	19.4	2.4
6c	6.0	1.5	2.7	2.7	—	—
6d	6.5	1.0	2.5	2.5	—	—
7	4.0	2.3	1.8	—	—	3.5

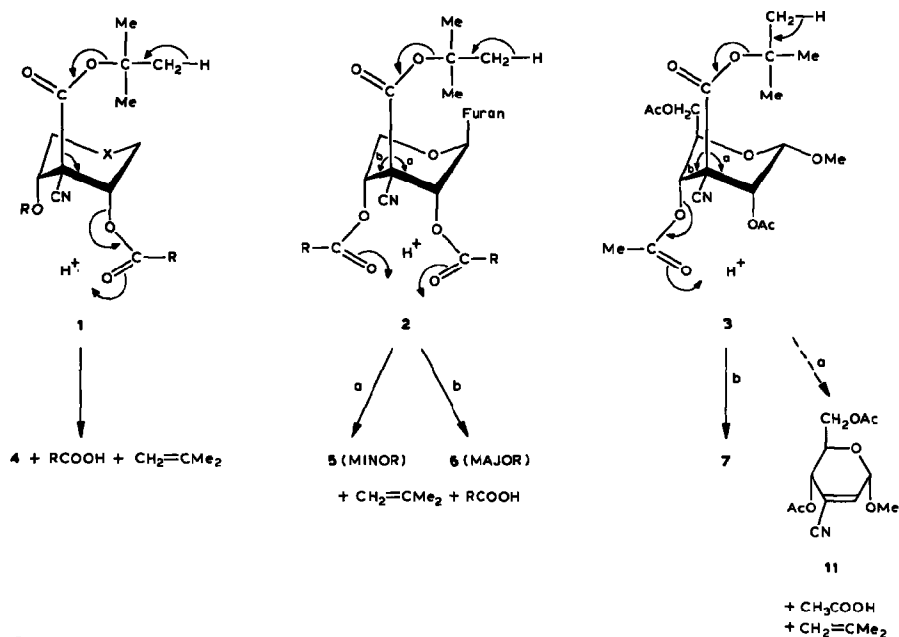
^aA series of decoupling experiments enabled assignments to be made.

(see Table IV) (quasi-axial-equatorial) accord with the ${}^0H_1(D)$ conformation **10**. This conformational preference is determined by the anomeric effect¹⁶ in competition with the allylic effect⁶.

The ^{13}C -n.m.r. data (see Experimental) of **4a-d**, **5a-d**, **6a-d**, and **7** accord with the assigned structures.

Both **5a** and **6a** were unaffected on treatment with acetic anhydride-acid.

Scheme 1 shows the *trans*-elimination yielding the products 4–7. The *meso* compounds **1a–d** yielded the racemic products **4a–d**. However, *trans*-elimination in the chiral compounds **2a–d** yielded **5a–d** and **6a–d** as minor and major products,



Scheme 1

respectively. The chiral compound **3** gave only one of the two possible isomers **7**. This finding accords with results⁷ on similar reactions of nitro compounds, which were ascribed to the electron-withdrawing effect of the glycosidic center and to the A^{1,2} effect¹⁷ of the nitro group. For the cyano compounds, only the first of these effects may be operative.

The structures of the products of the elimination reaction **4**–**7** confirmed the configurations at C-4 (**1a**–**d**) and C-3 (**2a**–**d**, **3**) that were previously tentatively assigned³.

EXPERIMENTAL

The general methods have been described¹⁸. N.m.r. spectra (¹H at 80 and 200 MHz, ¹³C at 20 MHz) were obtained with Bruker WP-80-SY and WP-200 spectrometers. Mass spectra (70 eV) were recorded with an AEI MS/30-VG spectrometer.

Reaction of 1a–d, 2a–d, and 3 with acetic anhydride and toluene-p-sulfonic acid. — A mixture of **1a**–**d**, **2a**–**d**, or **3**, acetic anhydride, and toluene-*p*-sulfonic acid (<5 mg) was stirred and boiled under reflux. Chloroform (75 mL) was added, and the mixture was washed with saturated aqueous NaHCO₃ (5 × 50 mL) and then water (25 mL), dried, filtered, and concentrated.

The following conditions were used:

Starting compound (g)	Ac ₂ O (mL)	Time (h)	Product (g, %)
1a (0.4)	6	3	4a (0.17, 83.3)
1b (2.5)	10	3	4b (0.43, 33.8)
1c (0.9)	10	3	4c (0.32, 67.5)
1d (1.61)	10	3	4d (0.62, 72.8)
2a (0.6)	7	3	5a (0.09, 22.5)
			6a (0.23, 57.5)
2b (0.7)	7	3	5b (0.07, 15.5)
			6b (0.3, 66.5)
2c (1.2)	7	5	5c (0.1, 13.2)
			6c (0.39, 51.4)
2d (0.78)	7	3	5d (0.1, 20.9)
			6d (0.15, 31.4)
3 (0.7)	7	3	7 (0.35, 79.7)

(a) *With trans,trans-3,5-diacetoxy-r-4-tert-butoxycarbonyl-4-cyanotetrahydropyran*³ (**1a**). Column chromatography (1:2 hexane–ether) of the crude product gave 5-acetoxy-4-cyano-5,6-dihydro-2*H*-pyran (**4a**), isolated as a syrup; $\nu_{\max}^{\text{film}}$ 3035, 2225, 1738, 1642, 1224, 1134, 1095, 1041, and 860 cm⁻¹. N.m.r. data: ¹H, see Tables I and II; ¹³C, 169.84 (COO), 147.42 (C-3), 115.78 (CN), 110.19 (C-4), 66.74, 64.70 (C-2,6), 63.01 (C-5), and 20.54 (MeCOO). Mass spectrum: *m/z* 168 (M⁺ +1), 167 (M⁺), 140 (M⁺ – CNH), 139 (M⁺ – CNH, H), 108 (M⁺ – C₂H₄O₂), and 98 (M⁺ – CN, C₂H₃O₂). (Found: C, 57.60; H, 5.71; N, 8.10. C₈H₉NO₃ calc.: C, 57.47; H, 5.42; N, 8.38%).

(b) With *trans,trans*-3,5-dibenzoyloxy-*r*-4-*tert*-butoxycarbonyl-4-cyanotetrahydropyran³ (**1b**). Column chromatography (1:2 hexane-ether) of the crude product gave 5-benzoyloxy-4-cyano-5,6-dihydro-2*H*-pyran (**4b**), m.p. 80–81° (from hexane-ether); $\nu_{\max}^{\text{KBr}}$ 3064, 2225, 1713, 1650, 1602, 1286, 1262, 1106, 1036, 861, 819, and 707 cm⁻¹. N.m.r. data: ¹H, see Tables I and II; ¹³C, δ 165.64 (COO), 147.58 (C-3), 133.54, 129.88, 129.07, 128.47 (C₆H₅COO), 115.94 (CN), 110.51 (C-4), 66.95, 64.86, and 63.75 (C-2,5,6). (Found: C, 68.02; H, 4.96; N, 5.86. C₁₃H₁₁NO₃ calc.: C, 68.11; H, 4.83; N, 6.11%).

(c) With *trans,trans*-3,5-diacetoxy-*r*-4-*tert*-butoxycarbonyl-4-cyanotetrahydrothiopyran³ (**1c**). Column chromatography (2:1 hexane-ether) of the crude product gave 5-acetoxy-4-cyano-5,6-dihydro-2*H*-thiopyran (**4c**), isolated as a syrup; $\nu_{\max}^{\text{film}}$ 3040, 2227, 1742, 1637, 1224, 1030, 990, 923, and 803 cm⁻¹. N.m.r. data: ¹H, see Tables I and II; ¹³C, δ 169.33 (COO), 145.32 (C-3), 116.30, 114.33 (CN, C-4), 63.75 (C-5), 27.94, 25.26 (C-2,6), and 20.36 (MeCOO). Mass spectrum: *m/z* 184 (M⁺ + 1), 183 (M⁺), 140 (M⁺ - C₂H₃O), 124 (M⁺ - C₂H₃O₂), 123 (M⁺ - C₂H₄O₂), and 122 (M⁺ - C₂H₄O₂ - H) (Found: C, 52.70; H, 5.17; N, 7.47. C₈H₆NO₂S calc.: C, 52.44; H, 4.95; N, 7.64%).

(d) With *trans,trans*-3,5-dibenzoyloxy-*r*-4-*tert*-butoxycarbonyl-4-cyanotetrahydrothiopyran³ (**1d**). Column chromatography (3:1 hexane-ether) of the crude product gave 5-benzoyloxy-4-cyano-5,6-dihydro-2*H*-thiopyran (**4d**), isolated as a syrup; $\nu_{\max}^{\text{film}}$ 3061, 2227, 1722, 1638, 1601, 1583, 1263, 1095, 891, 803, and 712 cm⁻¹. N.m.r. data: ¹H, see Tables I and II; ¹³C, δ 164.77 (COO), 145.50 (C-3), 133.08, 129.29, 128.64, 128.06 (C₆H₅COO), 116.35, 114.08 (CN, C-4), 64.39 (C-5), 27.86 and 25.16 (C-2,6). Mass spectrum: *m/z* 246 (M⁺ + 1), 245 (M⁺), 140 (M⁺ - C₇H₅O), 124 (M⁺ - C₇H₆O₂), 123 (M⁺ - C₇H₆O₂ - H), 105 (M⁺ - C₆H₆NOS) (Found: C, 63.85; H, 4.60; N, 5.92. C₁₃H₁₁NO₂S calc.: C, 63.65; H, 4.52; N, 5.71%).

(e) With 5-(2,4-di-*O*-acetyl-3-*tert*-butoxycarbonyl-3-cyano-3-deoxy- β -D-xylo-pentopyranosyl)-3-ethoxycarbonyl-2-methylfuran³ (**2a**). Column chromatography (1:1 hexane-ether) of the crude product gave, first, 5-(4-*O*-acetyl-3-cyano-2,3-dideoxy- β -D-glycero-pent-2-enopyranosyl)-3-ethoxycarbonyl-2-methylfuran (**5a**), m.p. 109–110° (from hexane), $[\alpha]_{\text{D}}^{25} +98^\circ$ (c 1, chloroform); $\nu_{\max}^{\text{KBr}}$ 3058, 3035, 2234, 1743, 1721, 1611, 1570, 1232, 1087, and 780 cm⁻¹. N.m.r. data: ¹H, see Tables III and IV; ¹³C, δ 169.98, 163.39 (2 COO), 160.64 (furan C-2), 146.08 (furan C-5), 144.95 (C-2), 115.60, 114.59, 111.82 (CN, C-3, furan C-3), 111.82 (furan C-4), 67.74, 62.65 (C-1,4), 63.44 (C-5), 60.43 (CH₃CH₂O), 20.76 (MeCOO), 14.35, and 13.83 (furan Me, CH₃CH₂O) (Found: C, 59.92; H, 5.70; N, 4.22. C₁₆H₁₇NO₆ calc.: C, 60.18; H, 5.36; N, 4.38%).

Eluted second was 5-(2-*O*-acetyl-3-cyano-3,4-dideoxy- β -D-glycero-pent-3-enopyranosyl)-3-ethoxycarbonyl-2-methylfuran (**6a**), m.p. 99–100° (from hexane), $[\alpha]_{\text{D}}^{25} -123^\circ$ (c 1, chloroform); $\nu_{\max}^{\text{KBr}}$ 3065, 3037, 2224, 1748, 1719, 1640, 1616, 1572, 1247, 1230, 1090, 1054, and 768 cm⁻¹. N.m.r. data: ¹H, see Tables III and IV; ¹³C, δ 169.34, 163.38 (2 COO), 160.04 (furan C-2), 164.57 (furan C-5), 145.76 (C-4),

115.16, 114.43 (CN, furan C-3), 111.30 (C-3), 110.71 (furan C-4), 70.76, 63.70 (C-1,2), 63.93 (C-5), 60.25 (CH₃CH₂O), 20.48 (MeCOO), 14.25, and 13.72 (CH₃CH₂O, and furan Me) (Found: C, 60.20; H, 4.98; N, 4.35. C₁₆H₁₇NO₆ calc.: C, 60.18; H, 5.36; N, 4.38%).

(f) With 3-acetyl-5-(2,4-di-O-acetyl-3-tert-butoxycarbonyl-3-cyano-3-deoxy- β -D-xylo-pentopyranosyl)-2-methylfuran³ (**2b**). Column chromatography (2:1 hexane-ether) of the crude product gave, first, 3-acetyl-5-(4-O-acetyl-3-cyano-2,3-dideoxy- β -D-glycero-pent-2-enopyranosyl)-2-methylfuran (**5b**), isolated as a syrup, $[\alpha]_D^{15} +67^\circ$ (c 1, chloroform); $\nu_{\max}^{\text{film}}$ 3030, 2228, 1742, 1676, 1595, 1560, 1227, 1090, 1050, and 735 cm⁻¹. N.m.r. data: ¹H, see Tables III and IV; ¹³C, δ 193.44 (CO), 169.93 (COO), 159.75 (furan C-2), 146.19 (furan C-5), 144.84 (C-2), 122.18 (furan C-3), 115.53 (CN), 112.81 (C-3), 111.26 (furan C-4), 67.73, 62.54 (C-1,4), 63.49 (C-5), 29.04 (furan Ac), 20.69 (MeCOO), and 14.40 (furan Me). Mass spectrum: *m/z* 289 (M⁺), 230 (M⁺ - C₂H₃O₂), 229 (M⁺ - C₂H₄O₂), 228 (M⁺ - C₂H₄O₂ - H), 214 (M⁺ - C₂H₄O₂ - CH₃), 202 (M⁺ - C₂H₄O₂ - CNH), and 186 (M⁺ - C₂H₄O₂ - C₂H₃O) (Found: C, 62.10; H, 5.43; N, 5.05. C₁₅H₁₅NO₅ calc.: C, 62.27; H, 5.22; N, 4.84%).

Eluted second was 3-acetyl-5-(2-O-acetyl-3-cyano-3,4-dideoxy-D-glycero-pent-3-enopyranosyl)-2-methylfuran (**6b**), isolated as a syrup, $[\alpha]_D^{15} -104^\circ$ (c 1, chloroform); $\nu_{\max}^{\text{film}}$ 3060, 3010, 2227, 1749, 1677, 1605, 1565, 1222, 1119, 1044, 949, 831, and 756 cm⁻¹. N.m.r. data: ¹H, see Tables III and IV; ¹³C, δ 193.24 (CO), 169.16 (COO), 158.94 (furan C-2), 146.84 (furan C-5), 145.78 (C-4), 121.86 (furan C-3), 114.95 (CN), 110.23 (C-3), 110.01 (furan C-4), 70.47, 63.45 (C-1,2), 63.98 (C-5), 28.75 (furan Ac), 20.24 (MeCOO), and 14.06 (furan Me). Mass spectrum: *m/z* 290 (M⁺ + 1), 289 (M⁺), 274 (M⁺ - CH₃), 230 (M⁺ - C₂H₃O₂), 229 (M⁺ - C₂H₄O₂), 228 (M⁺ - C₂H₄O₂ - H), 214 (M⁺ - C₂H₄O₂ - CH₃), 202 (M⁺ - C₂H₄O₂ - CNH), and 186 (M⁺ - C₂H₄O₂ - C₂H₃O) (Found: C, 62.58; H, 5.45; N, 4.50. C₁₅H₁₅NO₅ calc.: C, 62.27; H, 5.22; N, 4.84%).

(g) With 5-(2,4-di-O-benzoyl-3-tert-butoxycarbonyl-3-cyano-3-deoxy- β -D-xylo-pentopyranosyl)-3-ethoxycarbonyl-2-methylfuran³ (**2c**). Column chromatography (1:1 hexane-ether) of the crude product gave, first, 5-(4-O-benzoyl-3-cyano-2,3-dideoxy- β -D-glycero-pent-2-enopyranosyl)-3-ethoxycarbonyl-2-methylfuran (**5c**), isolated as a syrup, $[\alpha]_D^{15} +61^\circ$ (c 1, chloroform); $\nu_{\max}^{\text{film}}$ 3040, 2229, 1717, 1606, 1592, 1263, 1232, 1086, 780, and 712 cm⁻¹. N.m.r. data: ¹H, see Tables III and IV; ¹³C, δ 165.64, 163.37 (2 COO), 160.57 (furan C-2), 146.13 (furan C-5), 145.14 (C-2), 133.64, 129.98, 129.0, 128.52 (C₆H₅COO), 115.59, 114.55, 112.87 (CN, furan C-3, C-3), 111.76 (furan C-4), 67.83, 63.22 (C-1,4), 63.62 (C-5), 60.38 (CH₃CH₂O), 14.29, and 13.78 (furan Me, CH₃CH₂O). Mass spectrum: *m/z* 381 (M⁺), 336 (M⁺ - C₂H₅O), 276 (M⁺ - C₇H₅O), 260 (M⁺ - C₇H₆O₂), 259 (M⁺ - C₇H₆O₂ - H), 230 (M⁺ - C₂H₅O - C₇H₆O), 214 (M⁺ - C₇H₆O₂ - C₂H₅O), and 186 (C₁₁H₆O₃) (Found: C, 66.39; H, 5.22; N, 3.54. C₂₁H₁₉NO₆ calc.: C, 66.13; H, 5.02; N, 3.67%).

Eluted second was 5-(2-O-benzoyl-3-cyano-3,4-dideoxy- β -D-glycero-pent-3-

enopyranosyl)-3-ethoxycarbonyl-2-methylfuran (**6c**), isolated as a syrup, $[\alpha]_D^{15} -130^\circ$ (c 1, chloroform); $\nu_{\max}^{\text{film}}$ 3060, 2227, 1720, 1620, 1605, 1582, 1260, 1230, 1090, 916, 780, and 716 cm^{-1} . N.m.r. data: ^1H , see Tables III and IV; ^{13}C , δ 163.36 (2 COO), 159.99 (furan C-2), 146.60 (furan C-5), 146.20 (C-4), 133.61, 129.88, 128.77, 128.41 ($\text{C}_6\text{H}_5\text{COO}$), 115.25, 114.37 (CN, furan C-3), 110.71 (C-3, furan C-4), 70.86, 64.34 (C-1,2), 63.76 (C-5), 60.20 ($\text{CH}_3\text{CH}_2\text{O}$), 14.18, and 13.68 (furan Me, $\text{CH}_3\text{CH}_2\text{O}$). Mass spectrum: m/z 381 (M^+), 336 ($\text{M}^+ - \text{C}_2\text{H}_5\text{O}$), 260 ($\text{M}^+ - \text{C}_7\text{H}_6\text{O}_2$), 259 ($\text{M}^+ - \text{C}_7\text{H}_6\text{O}_2 - \text{H}$), 230 ($\text{M}^+ - \text{C}_2\text{H}_5\text{O} - \text{C}_7\text{H}_6\text{O}$), 214 ($\text{M}^+ - \text{C}_7\text{H}_6\text{O}_2 - \text{C}_2\text{H}_5\text{O}$), and 186 ($\text{C}_{11}\text{H}_6\text{O}_3$) (Found: C, 65.87; H, 5.29; N, 3.85. $\text{C}_{21}\text{H}_{19}\text{NO}_6$ calc.: C, 66.13; H, 5.02; N, 3.67%).

(h) With 3-acetyl-5-(2,4-di-O-benzoyl-3-tert-butoxycarbonyl-3-cyano-3-deoxy- β -D-xylo-pentopyranosyl)-2-methylfuran³ (**2d**). Column chromatography (3:1 hexane-ether) of the crude product gave, first, 3-acetyl-5-(4-O-benzoyl-3-cyano-2,3-dideoxy- β -D-glycero-pent-2-enopyranosyl)-2-methylfuran (**5d**), isolated as a syrup, $[\alpha]_D^{15} +87^\circ$ (c 1, chloroform); $\nu_{\max}^{\text{film}}$ 3062, 2230, 1722, 1676, 1600, 1560, 1264, 1230, 1106, 1087, 946, 737, and 713 cm^{-1} . N.m.r. data: ^1H , see Tables III, IV; ^{13}C , δ 193.38 (CO), 165.57 (COO), 159.70 (furan C-2), 146.24 (furan C-5), 145.03 (C-2), 133.66, 129.90, 128.88, 128.44 ($\text{C}_6\text{H}_5\text{COO}$), 122.16 (furan C-3), 115.54 (CN), 112.92 (C-3), 111.23 (furan C-4), 67.83, 63.11 (C-1,4), 63.69 (C-5), 29.03 (furan Ac), and 14.36 (furan Me). Mass spectrum: m/z 351 (M^+), 336 ($\text{M}^+ - \text{CH}_3$), 230 ($\text{M}^+ - \text{C}_7\text{H}_5\text{O}_2$), 229 ($\text{M}^+ - \text{C}_7\text{H}_6\text{O}_2$), 214 ($\text{M}^+ - \text{C}_7\text{H}_6\text{O}_2 - \text{CH}_3$), 202 ($\text{M}^+ - \text{C}_7\text{H}_6\text{O}_2 - \text{CNH}$), and 186 ($\text{C}_{11}\text{H}_6\text{O}_3$) (Found: C, 68.60; H, 5.07; N, 4.22. $\text{C}_{20}\text{H}_{17}\text{NO}_5$ calc.: C, 68.36; H, 4.87; N, 3.98%).

Eluted second was 3-acetyl-5-(2-O-benzoyl-3-cyano-3,4-dideoxy- β -D-glycero-pent-3-enopyranosyl)-2-methylfuran (**6d**), isolated as a syrup, $[\alpha]_D^{15} -124^\circ$ (c 1, chloroform); $\nu_{\max}^{\text{film}}$ 3070, 2227, 1725, 1676, 1601, 1563, 1263, 1108, 1072, 948, 828, 760, and 715 cm^{-1} . N.m.r. data: ^1H , see Tables III and IV; ^{13}C , δ 193.38 (CO), 165.07 (COO), 159.13 (furan C-2), 146.73 (furan C-5), 146.08 (C-4), 133.61, 129.78, 128.60, 128.40 ($\text{C}_6\text{H}_5\text{COO}$), 122.02 (furan C-3), 115.12 (CN), 111.17 (C-3), 110.19 (furan C-4), 70.82, 64.23 (C-1,2), 64.06 (C-5), 28.88 (furan Ac), and 14.09 (furan Me). Mass spectrum: m/z 352 ($\text{M}^+ + 1$), 351 (M^+), 336 ($\text{M}^+ - \text{CH}_3$), 230 ($\text{M}^+ - \text{C}_7\text{H}_5\text{O}_2$), 229 ($\text{M}^+ - \text{C}_7\text{H}_6\text{O}_2$), 214 ($\text{M}^+ - \text{C}_7\text{H}_6\text{O}_2 - \text{CH}_3$), 202 ($\text{M}^+ - \text{C}_7\text{H}_6\text{O}_2 - \text{CNH}$), and 186 ($\text{C}_{11}\text{H}_6\text{O}_3$) (Found: C, 68.65; H, 4.93; N, 4.25. $\text{C}_{20}\text{H}_{17}\text{NO}_5$ calc.: C, 68.36; H, 4.87; N, 3.98%).

(i) With methyl 2,4,6-tri-O-acetyl-3-tert-butoxycarbonyl-3-cyano-3-deoxy- α -D-glucopyranoside³ (**3**). Column chromatography (1:2 hexane-ether) of the crude product gave methyl 2,6-di-O-acetyl-3-cyano-3,4-dideoxy- α -D-erythro-hex-3-enopyranoside (**7**), isolated as a syrup, $[\alpha]_D^{15} +88^\circ$ (c 1, chloroform); $\nu_{\max}^{\text{film}}$ 3080, 2229, 1743, 1643, 1223, 1148, 1041, and 916 cm^{-1} . N.m.r. data: ^1H , see Tables III and IV; ^{13}C , δ 170.19, 169.66 (2 COO), 143.56 (C-4), 114.49 (CN), 111.13 (C-3), 94.86 (C-1), 66.49, 63.81 (C-2,5), 63.70 (C-6), 56.01 (MeO), 20.40 and 20.30 (2 MeCOO). Mass spectrum: m/z 270 ($\text{M}^+ + 1$), 269 (M^+), 238 ($\text{M}^+ - \text{CH}_3\text{O}$), 210 ($\text{M}^+ - \text{C}_2\text{H}_3\text{O}_2$), 196 ($\text{M}^+ - \text{C}_3\text{H}_5\text{O}_2$), 178 ($\text{M}^+ - \text{C}_2\text{H}_4\text{O}_2 - \text{CH}_3\text{O}$),

150 ($M^+ - C_2H_4O_2 - C_2H_3O_2$), and 136 ($M^+ - C_3H_5O_2 - C_2H_4O_2$) (Found: C, 53.85; H, 5.76; N, 5.32. $C_{12}H_{15}NO_6$ calc.: C, 53.52; H, 5.61; N, 5.20%).

REFERENCES

- 1 F. J. LOPEZ APARICIO, F. SANTOYO GONZALEZ, P. GARCIA MENDOZA, AND J. A. DOMINGUEZ MARTINEZ, *Carbohydr. Res.*, 147 (1986) 237–245.
- 2 F. J. LOPEZ APARICIO, F. SANTOYO GONZALEZ, P. GARCIA MENDOZA, AND J. A. DOMINGUEZ MARTINEZ, *An. Quim.*, in press.
- 3 F. J. LOPEZ APARICIO, F. SANTOYO GONZALEZ, P. GARCIA MENDOZA, F. HERNANDEZ MATEO, AND J. A. DOMINGUEZ MARTINEZ, *Carbohydr. Res.*, 152 (1986) 99–111.
- 4 L. J. BELLAMY, *The Infrared Spectra of Complex Molecules*, Methuen, London, 1954, pp. 31, 152, and 223.
- 5 F. W. McLAFFERTY, *Anal. Chem.*, 29 (1957) 1782–1789.
- 6 R. J. FERRIER AND G. H. SANKEY, *J. Chem. Soc., C*, (1966) 2345–2349.
- 7 H. H. BAER AND CHUNG-WAI CHIV, *Can. J. Chem.*, 52 (1974) 111–121.
- 8 R. C. LORD, T. C. ROUNDS, AND T. VEDA, *J. Chem. Phys.*, 57 (1972) 2572–2580.
- 9 J. R. DURIG, R. O. CARTER, AND L. A. CARREIRA, *J. Chem. Phys.*, 60 (1974) 3098–3103.
- 10 V. A. TIMOSHCHUK AND L. N. KULINKOVICH, *Carbohydr. Res.*, 145 (1985) 123–129.
- 11 P. BHATE, D. HORTON, AND W. PRIEBE, *Carbohydr. Res.*, 144 (1985) 331–337.
- 12 M. CHMIELEWSKI, J. JURCZACK, A. ZAMOJSKI, AND H. ADAMOWICZ, *Org. Magn. Reson.*, 20 (1982) 249–253.
- 13 E. W. GARBISCH, *J. Am. Chem. Soc.*, 86 (1964) 5561–5564.
- 14 N. S. ZEFIROV, V. S. BLAGOVESHCHENSKY, I. V. KAZIMIRCHIK, AND N. S. SUROVA, *Tetrahedron.*, 27 (1971) 3111–3118.
- 15 W. CAMERON, D. G. I. KINOSTON, N. SHEPPARD, AND LORD TODD, *J. Chem. Soc.*, (1964) 51–61.
- 16 R. U. LEMIEUX AND N. J. CHÜ, *Abstr. Pap. Am. Chem. Soc. Meet.*, 133 (1958) 31N.
- 17 F. JOHNSON AND S. K. MALHOTRA, *J. Am. Chem. Soc.*, 87 (1965) 5492–5493.
- 18 F. J. LOPEZ APARICIO, F. ZORRILLA BENITEZ, P. GARCIA MENDOZA, AND F. SANTOYO GONZALEZ, *Carbohydr. Res.*, 135 (1985) 303–311.